shown in Table 2 in accordance with the atomnumbering scheme shown in Fig. 1 (I) and Fig. 2 (II).*

Related literature. The inclination angles between the least-squares planes of the carbamate and phenyl groups, 86.3 (I) and 82.5° (II), differ from related molecules including dimethyl 4,4'-methylenebis-(phenylcarbamate) (Gardner & Blackwell, 1980) and bis(4-hydroxybutyl)-4,4'-methylenebis(phenyl carbamate) (Forcier & Blackwell, 1981). A structure reported by Born, Hocker, Paulus & Wolfel, (1981) shows a number of similarities.

We thank Dr Robert Saxon of American Cyanamide Company for providing the samples of p-TMXDI (Singh, Chang & Forgione, 1984). JPJ wishes to thank MTL and the US Army for an IPA fellowship and support of this work. JPJ also wishes to thank Dr E. M. Holt, Oklahoma State University, for her assistance and advice during the early stages of this work.

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The Structures of Dimethyl N, N'-trans-1,4-Cyclohexanedicarbamate (I) and Diethyl N, N'-trans-1,4-Cyclohexanedicarbamate (II)

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Abstract. (I) $C_{10}H_{18}N_2O_4$, $M_r = 230.27$, monoclinic, $P2_1/c$, a = 10.563 (2), b = 7.265 (1), c = 8.375 (1) Å, $\beta = 106.87$ (1)°, V = 615.0 Å³, Z = 2, $D_x =$ 1.243 g cm⁻³, Mo Ka, $\lambda = 0.71073$ Å, $\mu = 0.899$ cm⁻¹, F(000) = 248, room temperature. R = 0.058, 1041 independent observed reflections. (II) C₁₂H₂₂N₂O₄, $M_r = 258.32$, triclinic, PI, a = 8.251 (5), b =9.284 (5), c = 12.479 (7) Å, a = 69.30 (1), $\beta =$ 108.18 (1), $\gamma = 68.78$ (1)°, V = 719.7 Å³, Z = 2, D_x = 1.192 g cm⁻³, Mo Ka, $\lambda = 0.71073$ Å, $\mu =$

^{*} Lists of structure factors, anisotropic thermal parameters and H-atom parameters and packing diagrams have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 44737 (22 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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Table 1. Atomic coordinates $(\times 10^4)$ and equivalent isotropic thermal parameters $(Å^2)$ for (I) and (II) with

Table 2. Selected bond lengths (Å) and bond angles (°) for (I) and (II)

e.s.a. s in the least significant aight in parentheses					(I) $C_{10}H_{18}N_{2}O_{4}$ (II) $C_{12}H_{22}N_{2}O_{4}$			
	r	v	. 7	T T *	C(2)-C(3)	1.518 (3)	C(11)–C(12) 1.	504 (6)
(I)C H	NO	y	. 4	eq	C(2)–C(4)	1.523 (3)	C(11)–C(13) 1.4	490 (7)
$(1) = \frac{10^{14}18^{12}2}{4}$				C(1)–N(1)	1.326 (2)	C(15)–C(16) 1.4	422 (8)	
	8653 (2)	620 (2)	2820 (2)	48.3 (4)	C(1)–O(1)	1.338 (2)	C(21)–C(22) 1.4	493 (7)
0(2)	7599 (2)	542 (2)	68 (2)	52.6 (4)	C(1)-O(2)	1.218 (2)	C(21)-C(23) 1.	506 (7)
N(1)	7111 (2)	2673 (3)	1771 (2)	45.3 (4)	C(2) - N(1)	1.459 (2)	C(25)-C(26) 1.4	439 (8)
C(1)	7761 (2)	1231 (3)	1439 (2)	37.5 (4)	C(5) - O(1)	1.434 (3)	C(11)-N(1) 1.4	458 (6)
C(2)	6058 (2)	3615 (3)	519 (2)	39.6 (4)	., .,	• •	C(14) - N(1) = 1.3	312 (6)
C(3)	6380 (2)	5641 (4)	421 (3)	48.7 (5)			C(21) - N(2) = 1.4	452 (6)
C(4)	4736 (2)	3386 (4)	885 (3)	49.3 (5)		•	C(24) - N(2) = 1	325 (6)
C(5)	9436 (3)	-925 (4)	2614 (4)	67.9 (7)			C(14) = O(11) 1.	365 (6)
m c u	NO						C(14) = O(12) 1	189 (5)
$(11) C_{12} T$	221 204		••••				C(15) = O(11) 1.4	472 (6)
0(11)	1473 (4)	4246 (4)	2822 (3)	5.9 (1)			C(24) = O(21) 1.	347 (6)
O(12)	1220 (5)	2687 (4)	1822 (4)	8.7 (1)			C(24) = O(22) 1.	215 (5)
O(21)	3486 (4)	1146 (4)	6823 (3)	5.6(1)			C(25) = O(21) 1.	124 (6)
O(22)	3667 (5)	2977 (4)	7598 (3)	7.7 (1)			C(23) = O(21)	+2+ (0)
N(1)	-1522 (5)	4914 (4)	1491 (4)	5+3 (1)	C(3)-C(2)-C(4)	110.5 (2)	C(12)-C(11)-C(13)	110.2 (4)
N(2)	6464 (5)	529 (4)	8148 (4)	5.4 (1)	N(1)-C(2)-C(3)	111.0 (2)	C(22) - C(21) - C(23)	110.4 (4)
C(11)	-3001 (6)	4743 (5)	621 (4)	4.5 (2)	N(1) - C(2) - C(4)	110.9 (1)	N(1) - C(11) - C(12)	111.5 (5)
C(12)	-4804 (7)	6574 (6)	-517 (5)	5.4 (2)	N(1) - C(1) - O(1)	111-0 (l)	N(1) - C(11) - C(13)	110.4 (5)
C(13)	-3640 (8)	3661 (6)	1386 (5)	5.9 (2)	N(1)-C(1)-O(2)	125.6 (2)	N(2) - C(21) - C(22)	111.3 (5)
C(14)	422 (7)	3857 (6)	2003 (5)	5.2 (2)	O(1) - C(1) - O(2)	123.4 (2)	N(2)-C(21)-C(23)	110.6 (5)
C(15)	3674 (7)	2993 (7)	3559 (6)	8.5 (2)	C(1) - N(1) - C(2)	123.6 (2)	O(11) - C(15) - C(16)	108.4 (5)
C(16)	4571 (9)	3627 (9)	4300 (6)	10.8 (3)	C(1) = O(1) = C(5)	116.2 (2)	O(21) - C(25) - C(26)	109.4 (5)
C(21)	7926 (6)	697 (5)	9024 (5)	5.3 (2)			O(11) - C(14) - N(1)	110.7 (5)
C(22)	9458 (7)	698 (6)	8629 (5)	5.4 (2)			O(12) - C(14) - N(1)	126.5 (6)
C(23)	8939 (7)	-807 (6)	10443 (5)	5.9 (2)			O(21) - C(24) - N(2)	111.5 (5)
C(24)	4489 (6)	1646 (6)	7541 (5)	5.1 (2)			O(22) - C(24) - N(2)	124.9 (6)
C(25)	1353 (8)	2324 (7)	6033 (6)	8.0 (2)			O(11)-C(14)-O(12)	122.7 (5)
C(26)	545 (8)	1622 (9)	5295 (6)	9.4 (3)			O(21) - C(24) - O(22)	123.6 (5)
-		• • • • • • • • •				C(11) = N(1) = C(14)	122.4 (4)	
• $U_{eq} = \frac{1}{3}$ trace U_{ij} .							C(21) - N(2) - C(24)	123.8 (5)

 0.835 cm^{-1} , F(000) = 280, room temperature. R =0.066, 961 independent observed reflections. The angle between the least-sqares planes for the cyclohexane and carbamate groups is 87.5° in (I) and 82.3° in (II). In (I) a twofold axis relates the two halves of the molecule at $\frac{1}{2}$, $\frac{1}{2}$, 0 in the unit cell while (II) is centrosymmetric. The cyclohexane groups tend to stack in the ab plane nearly perpendicular to the c axis in (II) while in (I) the molecules are oriented in a cross-linked fashion.

Experimental. Clear colorless single crystals of dimensions $0.30 \times 0.30 \times 0.60$ mm (I) and $0.30 \times 0.40 \times 0.40$ 0.60 mm (II), grown by slow evaporation of a mixture of 5 g of *trans*-1,4-cyclohexane diisocyanate (CHDI) (Zentner, 1987) after being refluxed in 100 ml of absolute methanol (I) or ethanol (II), was mounted on a 0.20 mm glass fiber in each case with the needle axis nearly parallel to the φ axis of a Nicolet P3m autodiffractometer. Cell dimensions and space-group data were determined by standard methods (Calabrese, 1980; Corfield, Doedens & Ibers, 1967). For each compound: cell constants from $\pm 2\theta$ values of 15 reflections in the range $15-30^\circ$; scan range between 2θ settings 0.8 (I) or 1.2° (II) above and below $K\alpha_1$ and $K\alpha_2$; 2θ range $4.0 \le 2\theta \le 60^\circ$ (I), $4.0 \le 2\theta \le 50^\circ$ (II); four standard reflections monitored every 500 (I) or 1000 (II) reflections; data points corrected for Lorentzpolarization effects. (I): 1803 independent reflections;



C(14)-O(11)-C(15) C(24) - O(21) - C(25)

117.0 (4

Fig. 1. Molecular drawing of $C_{10}H_{18}N_2O_4$.



Fig. 2. Molecular drawing of C12H22N2O4.

index ranges $0 \le h \le 14$, $0 \le k \le 10$, $-11 \le l \le 11$. (II): 2673 independent reflections; index ranges $-9 \leq$ $h \le 9, -10 \le k \le 10, 0 \le l \le 13$. Each structure was solved by direct methods using a modified version of MULTAN80 (Main, Fiske, Hull, Lessinger, Germain, Declercq & Woolfson, 1980), SDP-Plus (B. A. Frenz & Associates Inc., 1984) and TEXRAY 234 (Molecular Structure Corporation, 1984). O, N and C atoms anisotropic, hydrogen atoms from $\Delta \rho$ map refined isotropically for three cycles and subsequently held fixed; $\sum w(|F_o| - |F_c|)^2$ minimized; $w = 4F_o^2/(\sigma F_o^2)^2$, $(\sigma F_o^2)^2 = [\tilde{s}^2(c + R^2B) + (pF_o^2)^2]/Lp^2$ and where p (ignorance factor used to downweight intense reflections) = 0.065 (I) or 0.050 (II) with $I > 3\sigma(I)$ (Corfield, Doedens & Ibers, 1967); no absorption corrections applied; features in final $\Delta \rho$ map ± 0.20 e Å⁻³; scattering factors and anomalousdispersion values were taken from standard sources (International Tables for X-ray Crystallography, 1974; Stewart, Davidson & Simpson, 1965). (I): R(109 variables) = 0.058 and wR = 0.087 for 1041 independent reflections; S = 0.06; $(\Delta/\sigma)_{max} = 0.001$. (II): R(163 variables) = 0.066 and wR = 0.068 for 961 independent reflections; S = 0.2; $(\Delta/\sigma)_{max} = 0.001$. Final atomic coordinates are listed in Table 1,* and bond lengths and angles in Table 2. Figs. 1 and 2 show perspective views of (I) and (II) with atomic numbering.

Related literature. The inclination angles between the least-squares planes of the cyclohexane and carbamate groups, 87.5 (I) and 82.3° (II), show similarities to structures reported by Jasinski, Desper, Zentner, Butcher & Day (1988). Other related molecules reported by Gardner & Blackwell (1980), Forcier & Blackwell (1981) and Born, Hocker, Paulus & Wolfel (1981) differ in this comparison but show a number of other similarities.

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Structure of Methyl 8,9-Epoxy-5,5-ethylenedioxy-7-oxo-11methoxytricyclo[7.2.1.0^{4,10}]dodec-3-ene-8-carboxylate

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Abstract. $C_{17}H_{20}O_7$, $M_r = 336\cdot34$, monoclinic, C2/c, $a = 15\cdot059$ (6), $b = 6\cdot888$ (2), $c = 30\cdot557$ (12) Å, β $= 94\cdot30$ (3)°, U = 3161 (2) Å³, Z = 8, $D_x =$ $1\cdot413 \text{ g cm}^{-3}$, λ (Mo $K\alpha$) = 0.71069 Å, $\mu = 1\cdot0 \text{ cm}^{-1}$, F(000) = 1424, T = 295 K, R = 0.059 for 1169 reflections with $I \ge 1\cdot5\sigma(I)$. The carboxylate moiety is

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^{*} Lists of structure factors, anisotropic thermal parameters, H-atom parameters and packing diagrams have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 44745 (21 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

coplanar with the adjacent C-O bond of the epoxy group. The bond lengths and angles are as expected.

Experimental. X-ray data for a plate-shaped transparent colourless crystal $(0.07 \times 0.45 \times 0.55 \text{ mm})$, glued on top of a glass fibre, were collected on an Enraf-Nonius CAD-4F diffractometer using Zr-filtered Mo Ka radiation. The crystals were found to be poorly

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